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Chapter 7 **Chemical** **Reactions** **Section 1 Review**

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Chemical

Reactions Review

ANSWER KEY 1.

synthesis $2\text{Sb} +$

$3\text{I}_2 \rightarrow 2\text{SbI}_3$ 2.

single

replacement 2Li

$+ 2\text{H}_2\text{O} \rightarrow 2\text{LiOH} +$

H_2 3.

decomposition

$2\text{AlCl}_3 \rightarrow 2\text{Al} +$

3Cl_2 4.

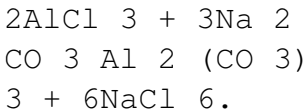
combustion C_6H_6

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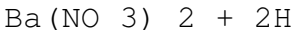
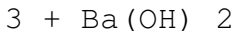
double

replacement



double

replacement 2HNO_3



7 . single

replacement ...

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**11.2 Types of
Chemical**

Reactions -

Evaluation 2016

Chemical

Reactions

Section 9.1

Reactions and

Equations In

your textbook,

read about

evidence of

chemical

reactions. For

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Key
each statement,
write yes if
evidence of a
chemical
reaction is
present. Write
no if there is
no evidence of a
chemical
reaction. 2. 4.
5. 6. 8. A
tomato smells
rotten. A
drinking glass

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breaks into
smaller pieces.
A piece of ...

Chemical Reactions Section Review Answer

Start studying
11.2 Types of
Chemical
Reactions. Learn
vocabulary,

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CHAPTER 8 REVIEW

Chemical

Equations and

Reactions

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SECTION 3 SHORT

ANSWER Answer

the following
questions in the
space provided.

1. List four
metals that will
not replace
hydrogen in an
acid. Choose
from Cu, Ag, Au,
Pt, Sb, Bi, and
Hg. 2. Consider
the metals iron

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Key
and silver, both
listed in Table
3 on page 286 of
the text. Which
one

18.1 section review

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Chemical

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Equations And
Key
Reactions

Section ...

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SECTION 3 SHORT

ANSWER Answer

the following
questions in the
space provided

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chemical

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equations and
reactions
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key. 1. List
four metals that
will not replace
hydrogen in an
acid.

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282 Chapter 9 •
Chemical
Reactions

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Section 99.1.1

Key
Figure 9.1 When adipoyl chloride in dichloromethane reacts with hexanediamine, nylon is formed. Nylon is used in many products, including carpeting, clothing, sports equipment, and

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tires.

Key

Objectives

Recognize

evidence of
chemical change.

Represent

chemical

reactions with
equations.

2.4 Chemical

Reactions

Section Review -

LPS

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Section 11.2

Review ... draw
concept maps
entitled "Types
of chemical
reactions." Have
them include all
the vocabulary
terms in the
concept map. Ask
them to explain
how ... Chemical
Reactions 331
Answers to...

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Key
Figure 11.5 The
reaction
produces bright,
white light.

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**Chapter 8 Review
Chemical**

**Equations And
Reactions
Section ...**

CHAPTER 8 REVIEW

Class Chemical
Equations and
Reactions

SECTION 2 SHORT

ANSWER Answer

the following
questions in the
space provided.

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Key

1. Match the equation type on the left to its representation on the right.

a synthesis
decomposition si
ngle-
displacement dou
ble-displacement
(a) $AX + BY \rightarrow AY + BX$
(b) $A + BX \rightarrow AX + B$

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8 Chemical

Equations and
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ANSWER Answer

the following
questions in the
space provided.

1. List four

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Key
metals that will
not replace
hydrogen in an
acid. Choose
from Cu, Ag, Au,
Pt, Sb, Bi, and
Hg. 2. Consider
the metals iron
and silver, both
listed in Table
3on page 286 of
...

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11.2 Types of

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Equations and Reactions

In a chemical reaction, reactants are changed into products.

_____11. The amount of matter present appears to change during a chemical reaction. 8 7 6
4 5 3 2 1 •

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Key

chemical

reaction •

reactants •

products •

chemical

property • law

of conservation

of mass CHEMICAL

REACTIONS

SECTION REVIEW

2.4

DESCRIBING

CHEMICAL

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Equations and

Reactions MIXED

REVIEW SHORT

ANSWER Answer

the following

questions in the

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space provided.

1. b A balanced chemical equation represents all the following except (a) experimentally established facts. (b) the mechanism by which reactants combine to form products.

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Answer Key:

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- Weebly

The actual knowledge about what products are produced in a chemical reaction is obtained by (a) inspecting the chemical equation; (b) ...
1, Vnte the

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Key
answers to the
following in the
... Chemical
Equations

Section 8.1 .

Name Types of
Chemical

Reactions Class

Date Section

Review

DIRECTIONS:

Write on the
line at the right
of each

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Key

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letter ...

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1 Chemical

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Chapter 8

SECTION 1 SHORT

ANSWER 1. a. d

b. a c. b d. f

e. e f. c 2.

8, 4, 9 3. a. 12

atoms b. 16

atoms c. 51

atoms d. 3 1024

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11.1 Describing Chemical Reactions Flashcards | Quizlet

Answer Key:

Chemical
Reactions
Review.

reactants on the
left, products
on the right.

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iodine, bromine,
chlorine,

fluorine,

oxygen,

nitrogen,

hydrogen. a

whole number in

front of an

element or

compound in a

chemical

equation; used

to balance the

equation.

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